

## Supplementary material

### ENVIRONMENTAL SCIENCE AND POLLUTION RESEARCH

#### Degradation of industrial surfactants by photocatalysis combined with ozonation

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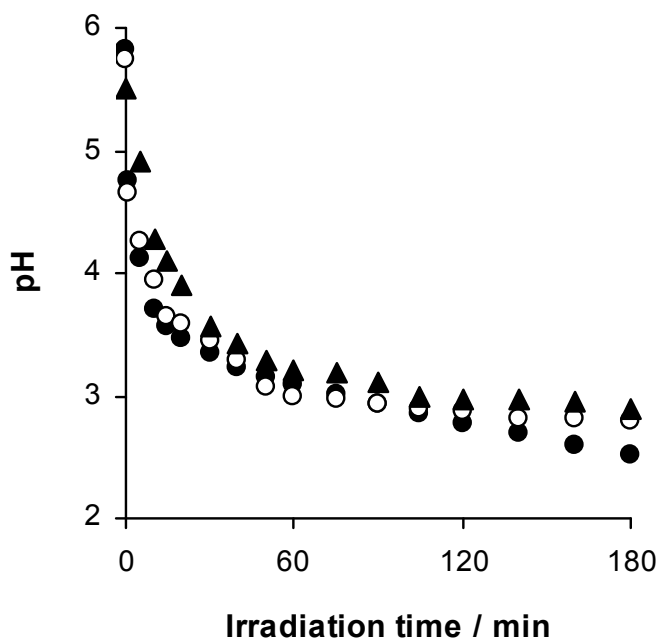
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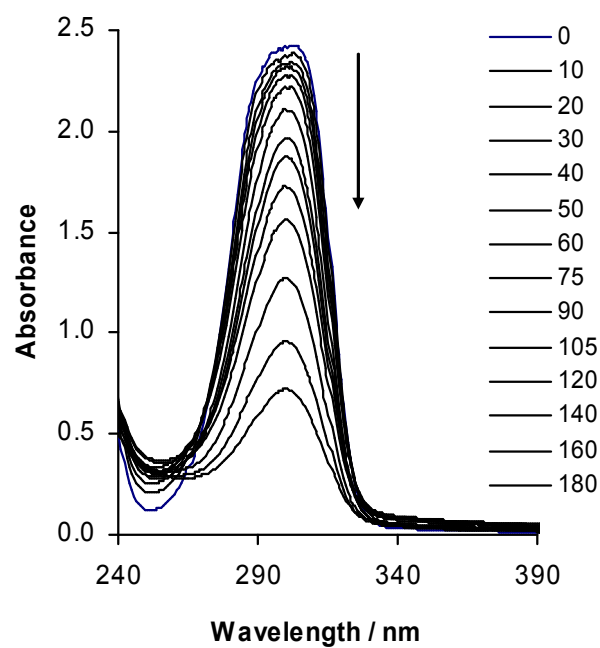
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**Fig. S1** Change of pH as functions of time in the system containing  $1 \times 10^{-3}$  M HBS during the treatment by various oxidation methods: (▲) O<sub>3</sub> + UV, (○) air+TiO<sub>2</sub> + UV, (●) O<sub>3</sub>+TiO<sub>2</sub> + UV. (TiO<sub>2</sub> content is  $1 \text{ g dm}^{-3}$ )



**Fig. S2** Change of the absorption spectrum of the bulk solution during the combined ( $\text{O}_3 + \text{TiO}_2 + \text{UV}$ ) treatment of the system containing  $1 \times 10^{-3}$  M DHBS and  $1 \text{ g dm}^{-3}$   $\text{TiO}_2$ . ( $\ell = 1.0 \text{ cm}$ ). The samples were taken at the indicated times in min